



Thermal-hydrologic-chemical Modeling for Experiment Comparison

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FEHM Model Development for WIPP Site Simulations

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Model Development Objectives

- **1.** Build confidence in the safety case
- 2. Validate model results on simple experiments
- 3. Support the planning, design, and interpretation of laboratory and field thermal tests
- 4. Optimize instrument placement and data collection in field testing with sensitivity analysis
- 5. Reduce uncertainty in simulated performance under full scale repository conditions
- 6. Code comparisons (FEHM, PFLOTRAN, TOUGH2, CODEBRIGHT)



FEHM Model Development for WIPP Site Simulations

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New code developments have allowed for additional physics to be represented in FEHM simulations, new experiments include:

- Thermal conductivity comparison to Bechthold (2004) equation (aka BAMBUSII)
 - Experiment and model design
 - Additional FEHM abilities
- Olivella experiment and comparison of FEHM results to CODE-BRIGHT and TOUGH2
 - Experiment
 - FEHM Modeling
 - Code Comparison
- Salt pan evaporation experiment
 - Mass change with relative humidity
 - FEHM code development for appropriate vapor pressure lowering



- Thermal conductivity of a material is an important control on how efficiently heat conducts in a system
- Complicated physics occurring around heat source in RoM salt which is controlled to some extent by thermal conductivity
- Goal is to use experimental and modeled data to lock down values of thermal conductivity for WIPP salt for future modeling



(from Jordan et al., 2015)



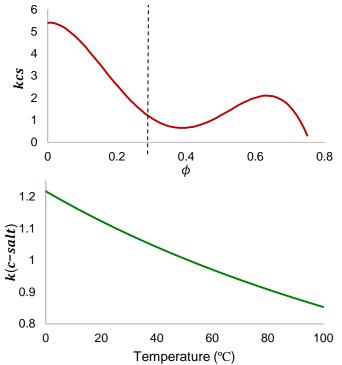
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- May need to improve thermal conductivity for WIPP RoM salt for improved simulation of salt behavior to a heat source
- Currently FEHM uses thermal conductivity calculated from porosity based on a relationship from the BAMBUSII study
- Relationship may not be appropriate for WIPP site salt determined from field experiments of drift-backfilled and crushed salt

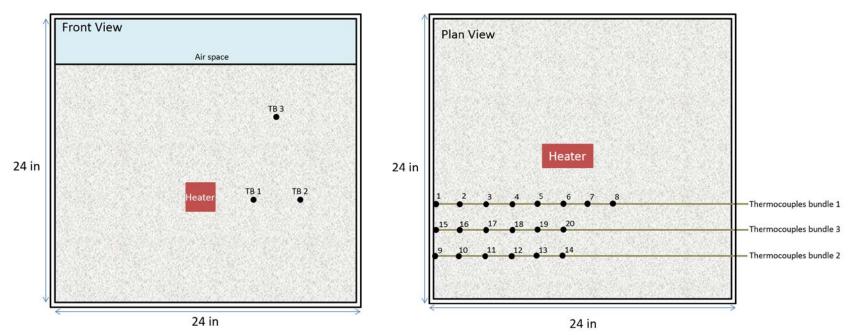
$$k_{cs}(\phi) = \begin{pmatrix} -270\phi^4 + 370\phi^3 + \cdots \\ 136\phi^2 + 1.5\phi + 5 \end{pmatrix} \cdot f$$

$$k_{c-salt}(T) = k_{cs}(\phi) \left(\frac{300}{T}\right)^{\gamma}$$

 $k_{cs} = thermal \ conductivity \ (crushed \ salt),$ $\phi = porosity, f = fitting \ factor,$ $T = temperature, \gamma = material \ constant$ (from Bechthold, 2004)

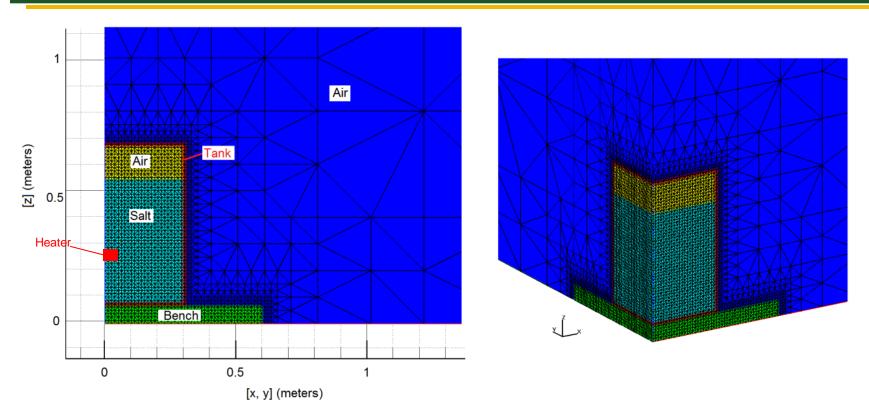






- Total mass of salt = 500 lb +- 20
- Moisture content = 0.2 wt. %
- 24" by 24" Plexiglas Box in laboratory
- Three thermocouple bundles with total of 20 sampling locations
- Tested three heater temperatures (80°C, 160°C, and 260°C)

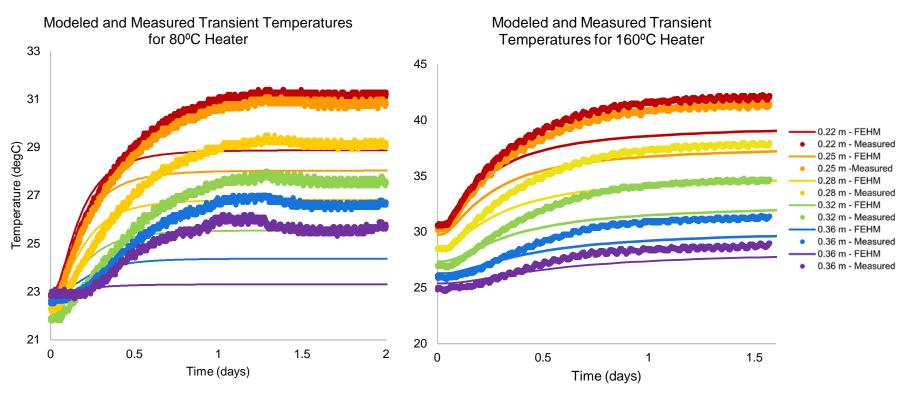




- Three dimensional model, quarter-space model includes RoM salt, air inside and outside of the box, Plexiglas walls, and lab bench
- Used BAMBUSII relationship for thermal conductivity
- Ran 3 simulations of heater temperature set to 80°C, 160°C, and 260°C



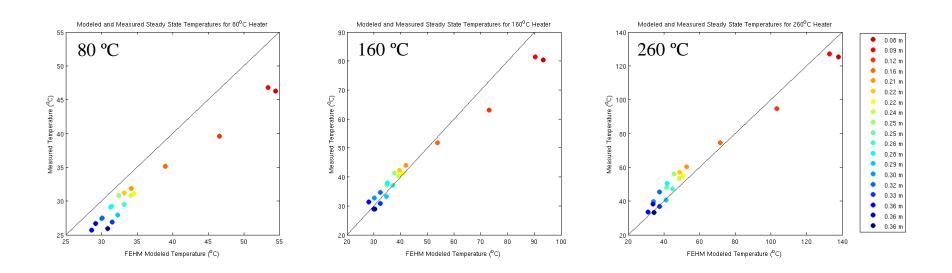
FEHM Model Development for WIPP Site Simulations



- BAMBUSII function of thermal conductivity:
- Over predicts time of heat equilibration for 80°C case, under predicts for 160°C case
- Under predicts steady-state temperature for both cases



FEHM Model Development for WIPP Site Simulations



- Steady state measured and modeled temperatures from salt box experiment show the thermal conductivity derived from BAMBUSII does not work well for all temperatures:
- Poor fit to 80 °C test
- Poor fit close to heater
- Gas tracer data may give clues to change in behavior for different temperatures



FEHM Model Development for WIPP Site Simulations

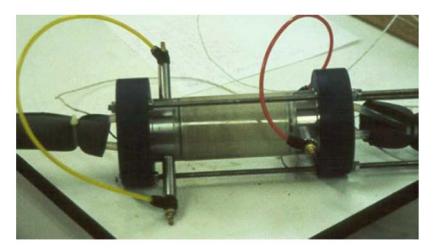
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The poor fit to the data indicates a new function is needed for WIPP-site simulations

- BAMBUSII determined salt based on different site with possibly different humidity, saturation, impurities in salt, and other environmental conditions
- We will use parameter optimization techniques to determine a new function for thermal conductivity to better fit for the RoM salt from WIPP site
 - PEST is a flexible parameter estimation software we can use to optimize a new thermal conductivity function to best reproduce the temperature changes for all 20 measurements for the three temperature experiments



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Sample	Cylindrical with horizontal axis
Diameter	50 mm
Length	100 mm
Initial porosity	30%
Initial saturation degree of brine	40%
Grain size	1–2 mm
Hot side temperature	85°C
Cold side temperature	5°C
Testing time	15 days

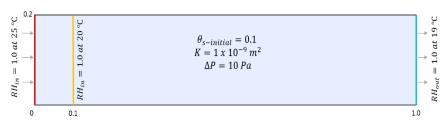
Goal is to develop new capabilities for FEHM for salt problems and validate FEHM against experimental data and other codes (CODE-BRIGHT, TOUGH2)

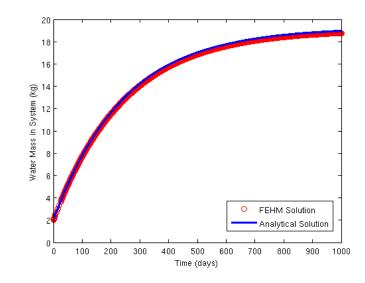
Canister experiment (Olivella et al., 2011)

- 0.1 m canister of salt, heated ends to 85°C and 5°C
- Characterized salt matrix properties at 4 times following heating
- Used CODE-BRIGHT to model system and compare experimental results

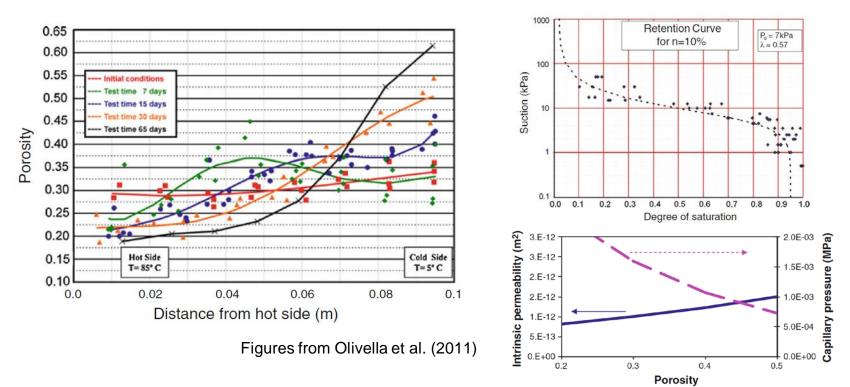


- Interested in porosity changes, brine migration due to large temperature gradient
- Equations used for modeling using CODE_BRIGHT (Olivella et al., 2011) are newly incorporated into FEHM salt modeling capabilities
 - Similarities to test problem ngas macro improvement
 - The outflow fluid mixture with constant pressure was corrected to include the correct mixture of air and water
 - Condensation and evaporation controlled by temperature gradient



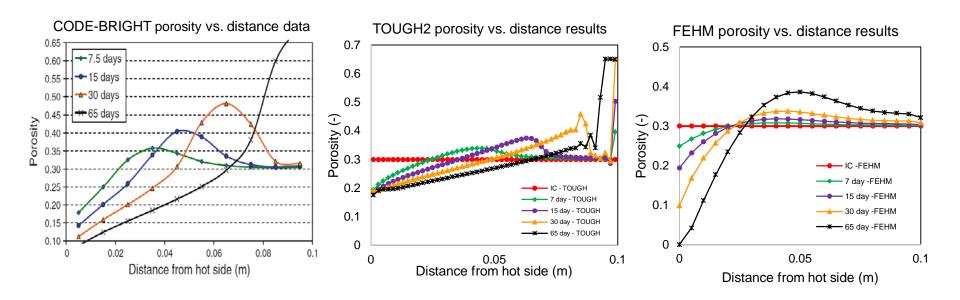






- FEHM improvements for simulations:
 - retention curves in FEHM runs to correspond with data from Olivella (2011)
 - salt module includes equations used in CODE-BRIGHT simulations
 - FEHM does not accommodate a relationship between capillary and porosity





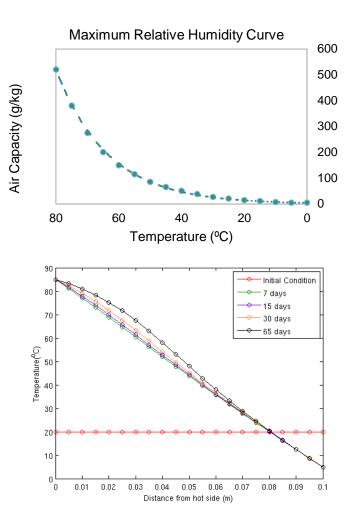
- All data show decrease in porosity on the hot side where evaporation is greatest, increased porosity on cold side due to condensation
- FEHM not producing porosity change on the cold side that other codes are replicating:



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What is driving the observed increase in porosity?

- Air holding capacity gradient not large enough to drive at cold end to drive such high condensation (dissolution of salt)
- Other simulation we ran with higher 'cold-end' temperatures do show high porosity



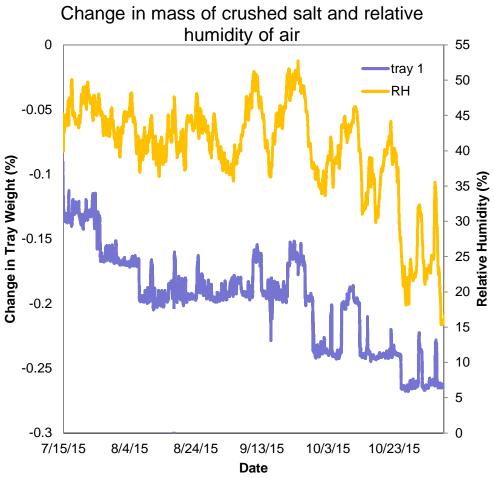


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Change in weight of crushed salt shows evaporation and condensation of water from the air

 Increased evaporation when relative humidity drops below ~40%



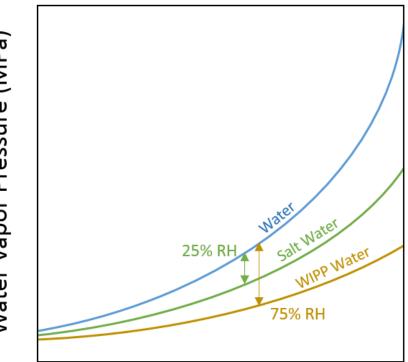




Pan Evaporation Experiment

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- Salt and other dissolved solids lower water vapor pressure, which increases retention of water in pores spaces
- No established relationship for WIPP site water
- Issues of FEHM using water vapor pressure - temperature relationship not representing the water at the WIPP site
- Performing long-term, on site experiment to determine water vapor pressure lowering for WIPP site
- Match model results to determine WIPP site relationship



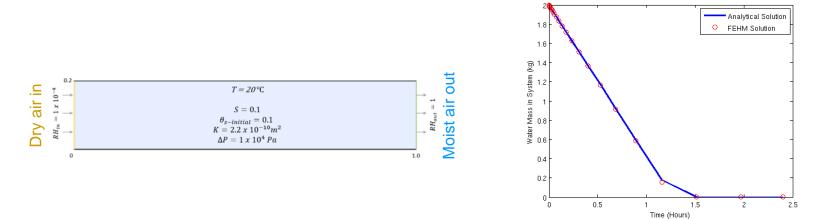
Temperature

Water Vapor Pressure (MPa)



Pan Evaporation Experiment

- The boundary conditions for air-water-heat physics are now available within the boun control statement
 - New keyword 'fxa' for flowing ngas mass fraction added. This keyword apportions incoming flow when the flow arises from a fixed pressure condition
 - The boun control statement allows multiple time changes for all boundary values so that experimental time series input can be used
- Improved the stability in other ngas boundary conditions so that boundary cells can dry-out or become fully saturated without causing extra iterations

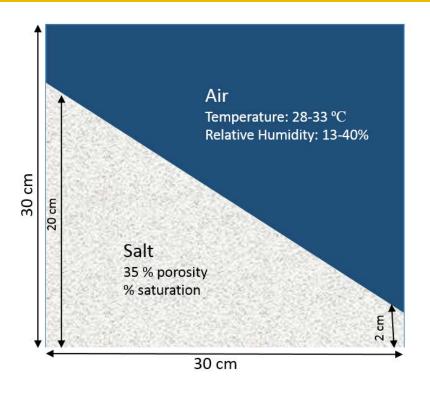




Pan Evaporation Experiment



- 2-D Radially symmetric model of mounded salt in a tray
- Time-variant boundary conditions of temperature and relative humidity (mass fraction of water in air)
- Water vapor pressure-lowering for pure salt water in use with FEHM salt capabilities



- Model domain of salt cone
- No-flow boundaries on all sides



Future Work

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Tackle challenges in code to better represent physics in experiments:

- Thermal conductivity experiment:
 - Determine function of thermal conductivity for WIPP salt
 - Add gas migration to model for measure of temperature-dependent diffusivity
- Olivella (2011) experiment with code comparison:
 - Why does experiment and other codes show large porosity change on cold end?
 - Why doesn't FEHM reproduce this result?
- Salt pan evaporation:
 - Determine water vapor pressure lowering relationship for WIPP-site salt